

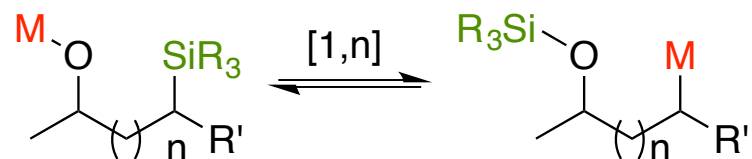
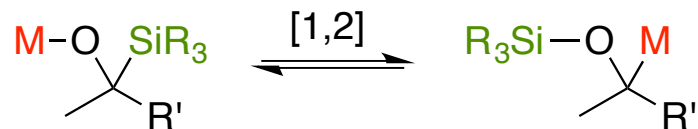
Brook Rearrangement as Trigger for Carbene Generation: Synthesis of Stereodefined and Fully Substituted Cyclobutenes

F. Ahang and Ilan Marek
J. Am. Chem. Soc. **2017**, *139*, 8364-8370.

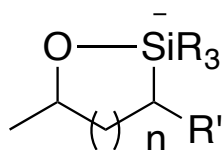
06/24/17

Mike Frasso

General Brook Rearrangement



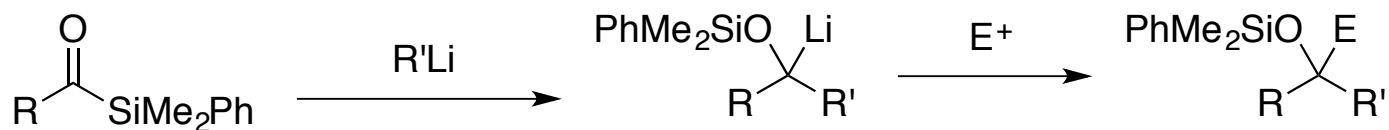
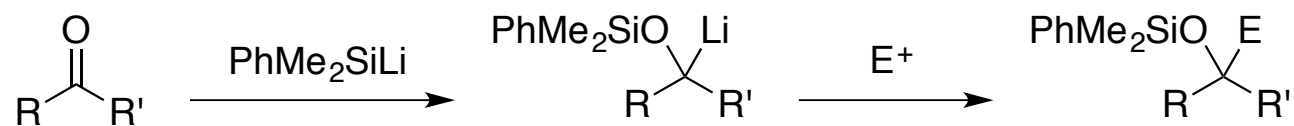
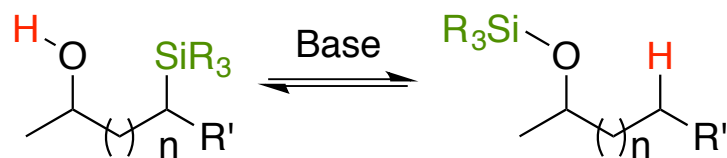
via



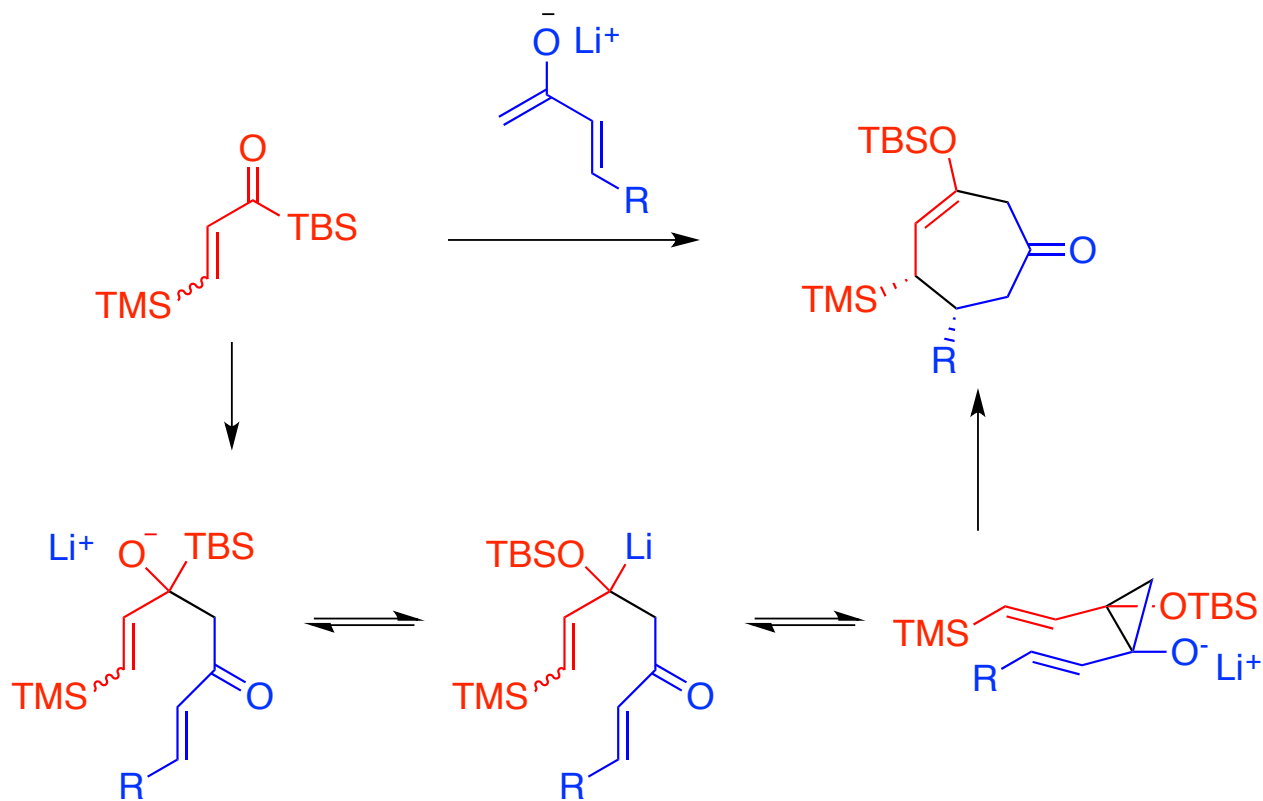
- If catalytic base used, stronger Si-O determines equilibrium
 - Si-O = 120-130 kcal/mol, Si-C = 75-85 kcal/mol
- If stoichiometric base used, anion stability determines equilibrium
 - R' = EWG, polar aprotic solvents, chelating agents, and M = Na, K (but not Li) all favor Si-O bond formation

Tetrahedron, **2001**, *57*, 2065-2084.

General Brook Rearrangement

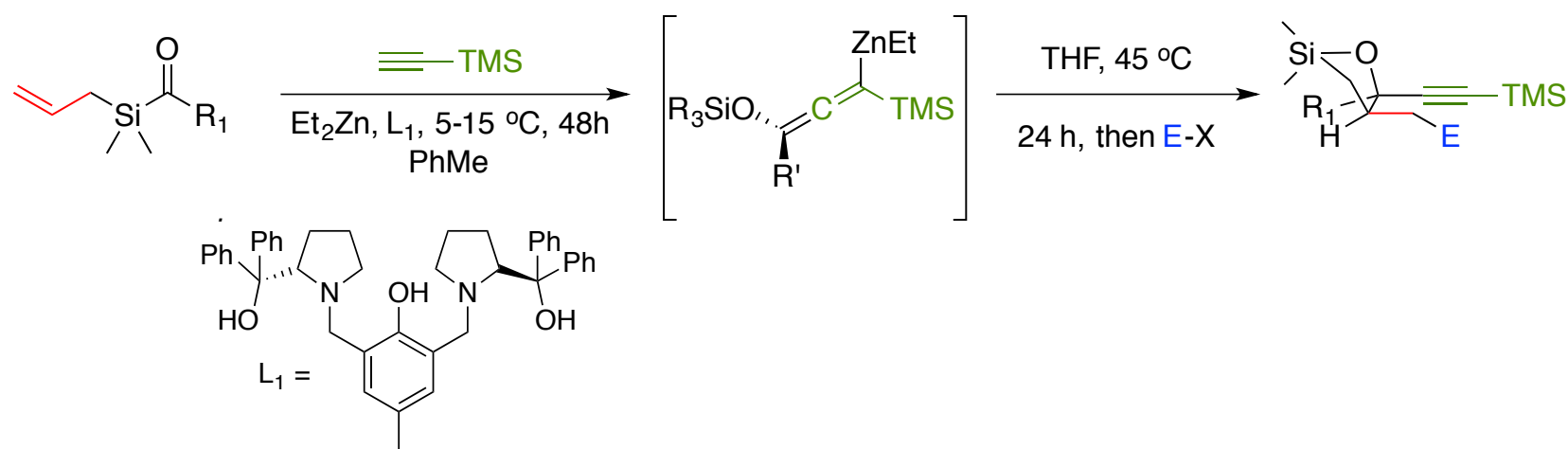


Brook Rearrangement: Formal [4+3]



Tetrahedron, **2001**, *57*, 2065-2084.
J. Am. Chem. Soc. **1998**, *120*, 4947.

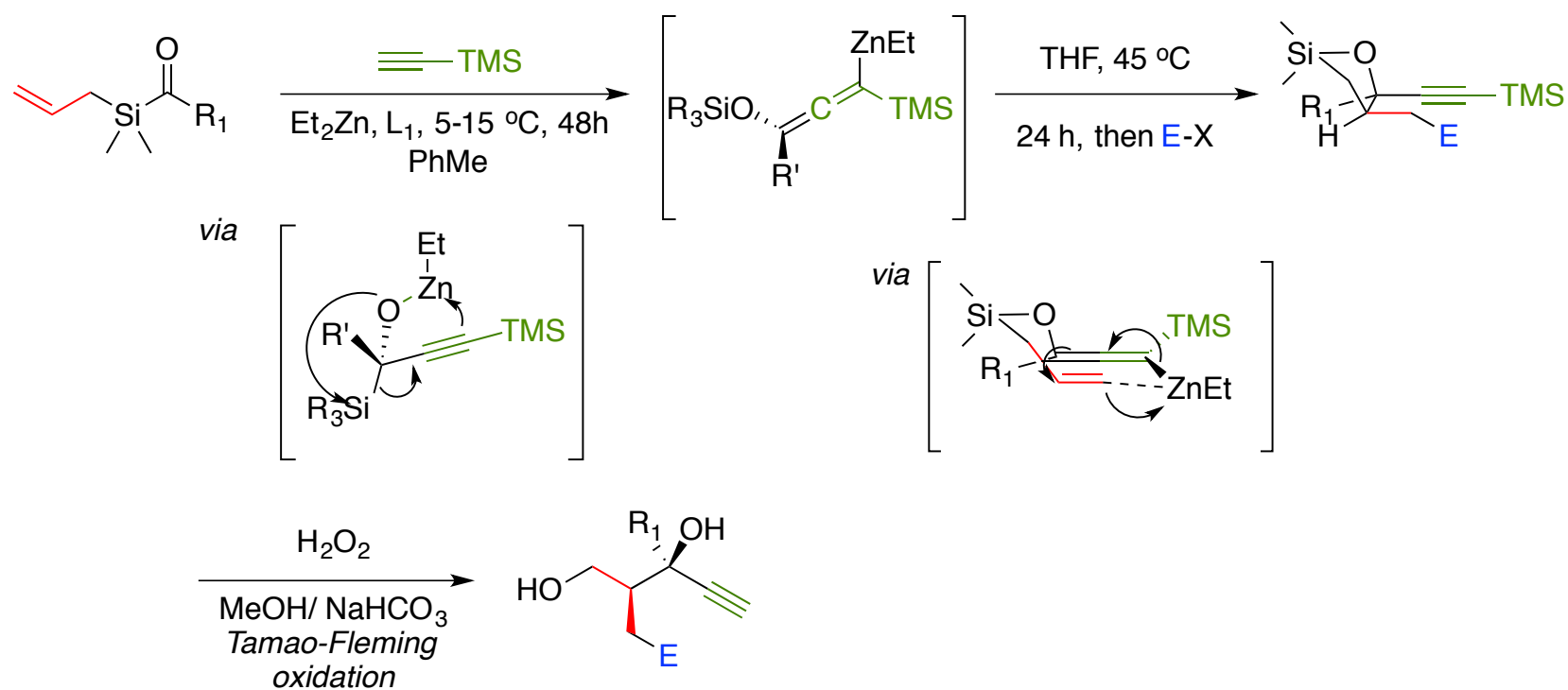
Allenyl-Zn Brook Rearrangement



- Ligand has no effect on stereochemistry after alkyne addition
- Believed to proceed directly to allenyl-Zn (configurationally stable)
 - No traditional 1,2-Brook rearrangement (alkyl-Zn not configurationally stable)

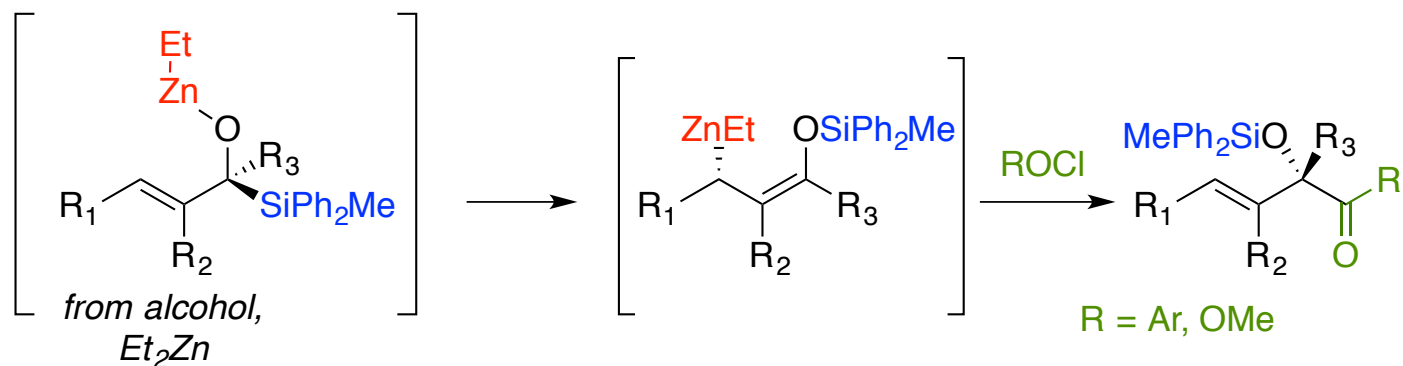
Angew. Chem. Int. Ed. **2013**, *52*, 13717–13721.

Allenyl-Zn Brook Rearrangement



Angew. Chem. Int. Ed. **2013**, *52*, 13717–13721.

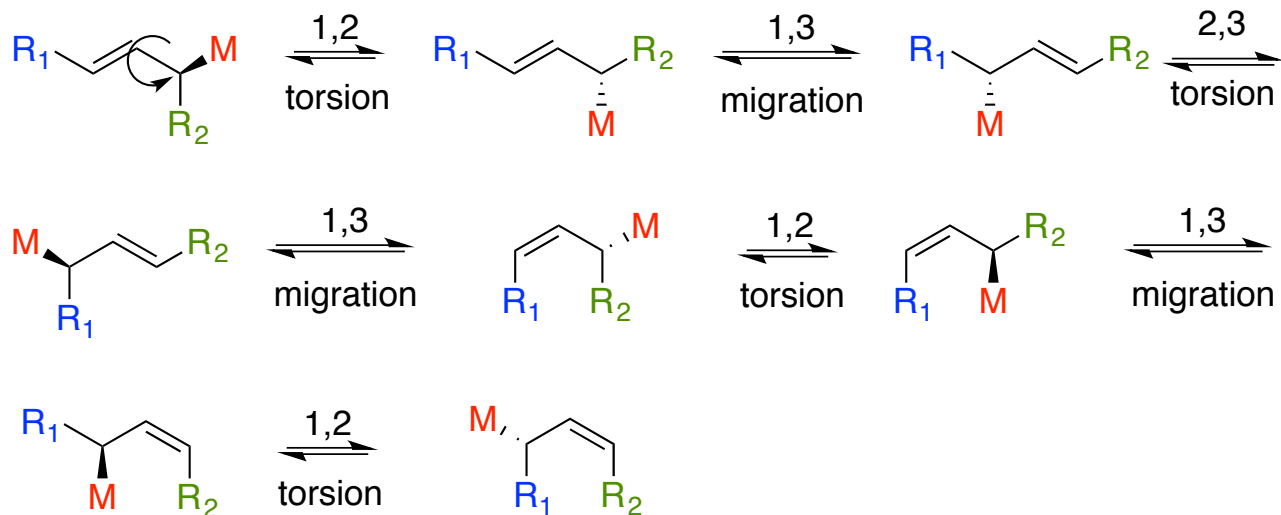
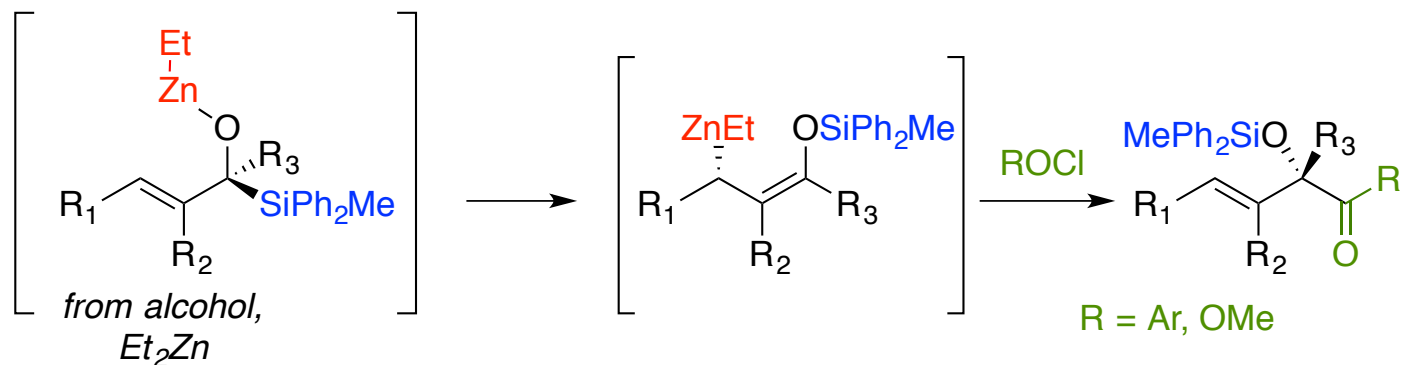
Allyl-Zn Brook Rearrangement



- Complete chirality transfer from starting alcohol
- 60-96% yield
- MeMgBr could also be used, but with decrease yield
- Smaller silyl groups (TMS) give *O*-acylation

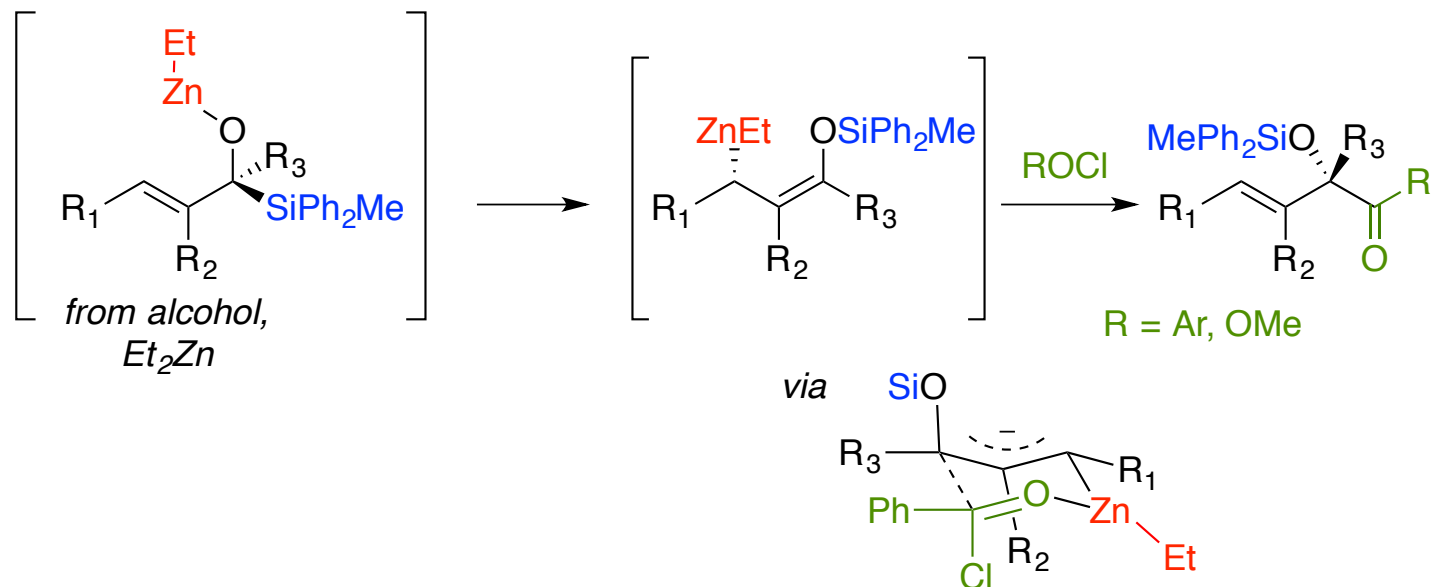
Angew. Chem. Int. Ed. **2016**, *55*, 6057–6061.

Allyl-Zn Brook Rearrangement



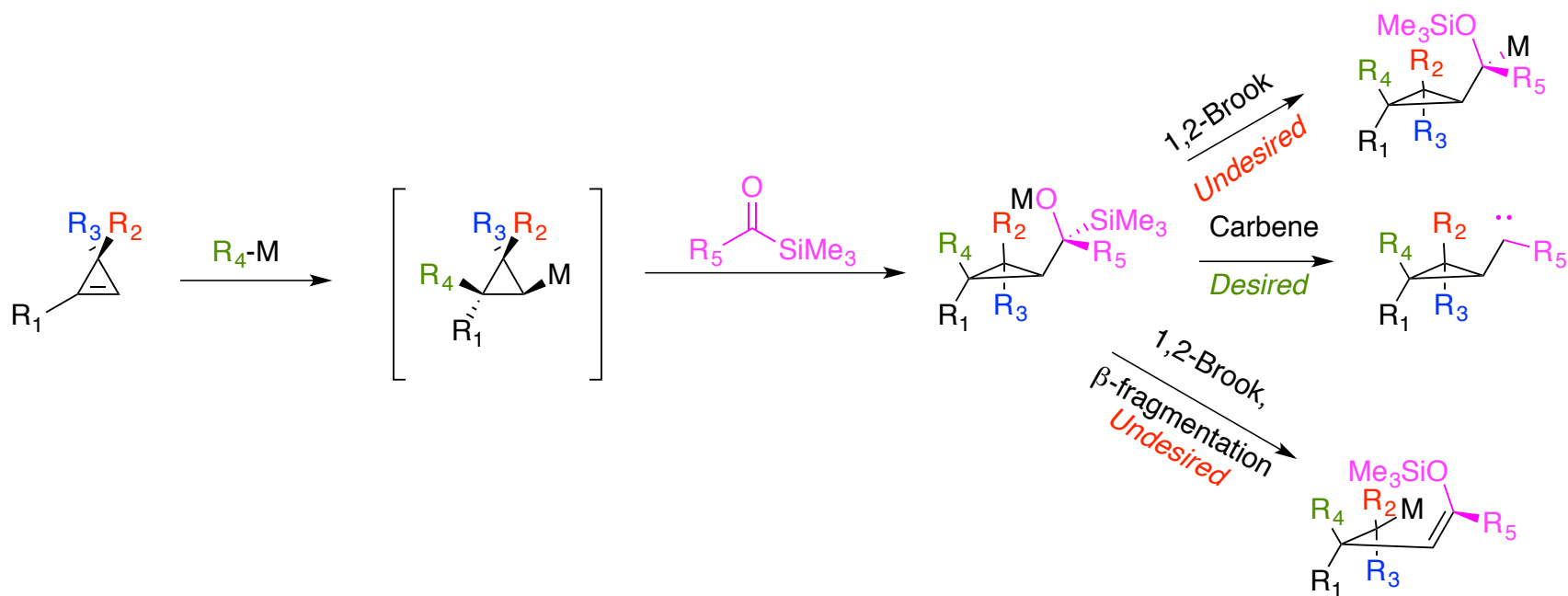
Angew. Chem. Int. Ed. **2016**, *55*, 6057–6061.

Allyl-Zn Brook Rearrangement



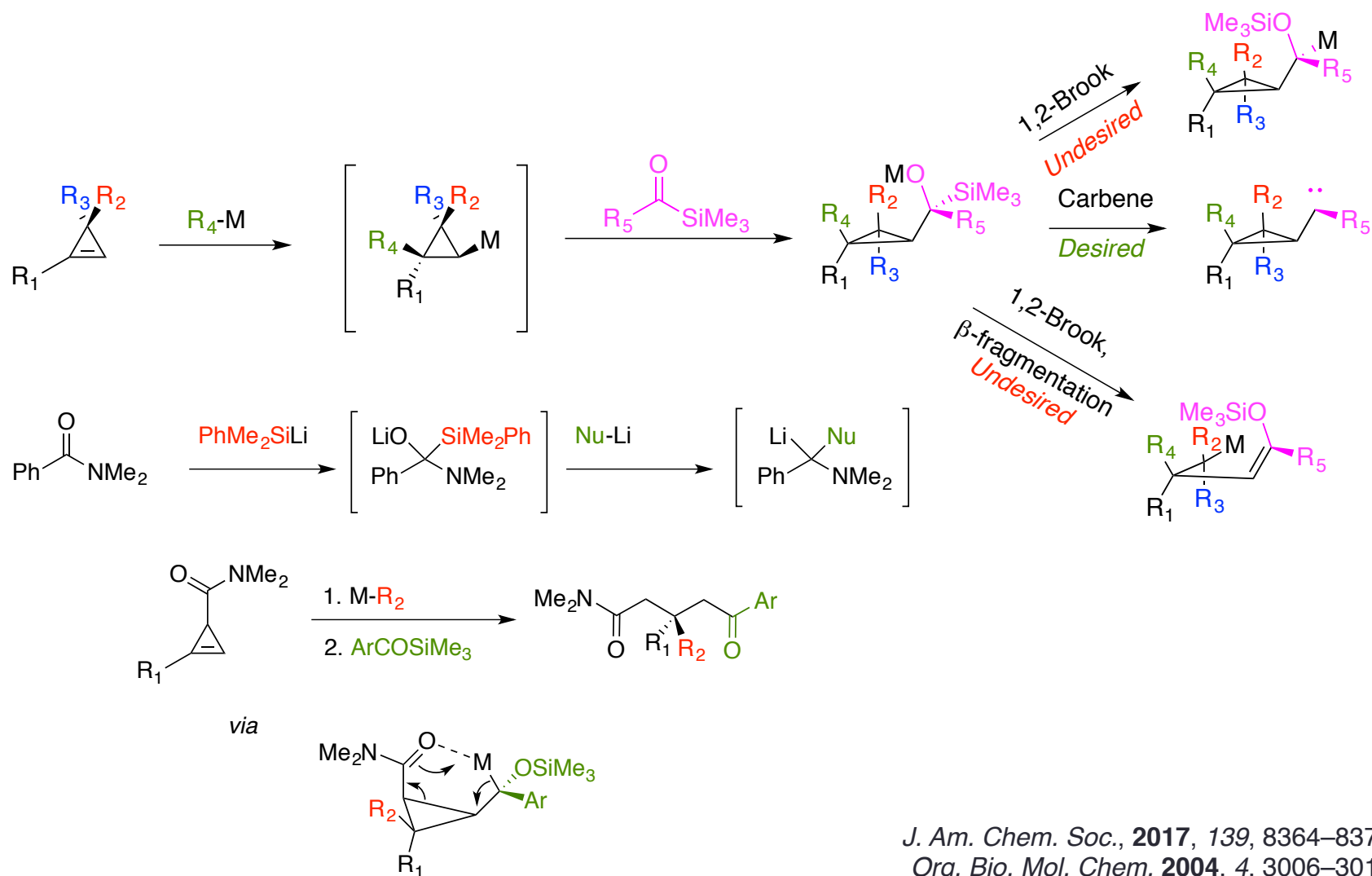
Angew. Chem. Int. Ed. **2016**, *55*, 6057–6061.

Title Work: Multiple Reaction Pathways



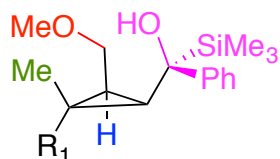
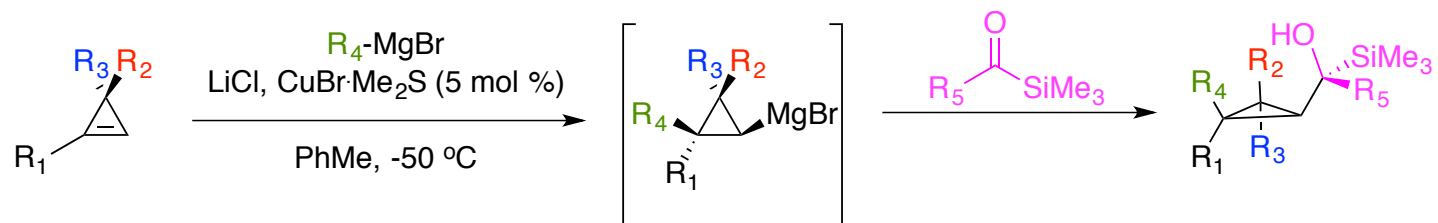
J. Am. Chem. Soc., **2017**, *139*, 8364–8370.
Org. Bio. Mol. Chem. **2004**, *4*, 3006–3017.
Angew. Chem. Int. Ed. **2016**, *55*, 714–718.

Title Work: Multiple Reaction Pathways

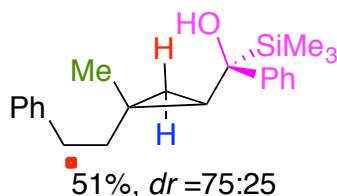


J. Am. Chem. Soc., **2017**, *139*, 8364–8370.
Org. Bio. Mol. Chem. **2004**, *4*, 3006–3017.
Angew. Chem. Int. Ed. **2016**, *55*, 714–718.

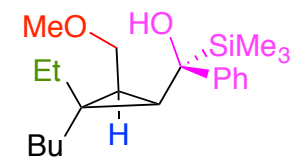
Title Work: Carbometallation/Acyl Silane Addition



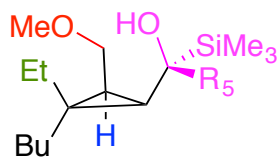
$R_1 = \text{Bu}$; 75%, $dr = 95:5$
 $R_1 = \text{Et}$; 78%, $dr = 95:5$
 $R_1 = \text{Hex}$; 72%, $dr = 95:5$
 $R_1 = \text{Ph}(\text{CH}_2)_2$; 64%, $dr = 98:2$
 $R_1 = \text{BnO}(\text{CH}_2)_2$; 71%, $dr = 98:2$



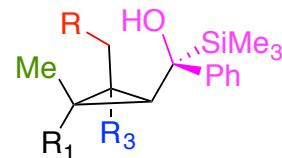
$R = \text{OBn}$; 65%, $dr = 98:2$
 $R = \text{OH}$; 84%, $dr = 98:2$



51%, $dr = 92:8$



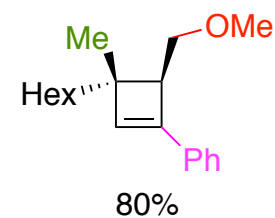
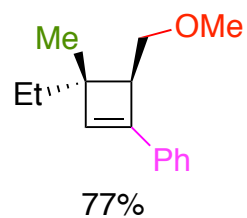
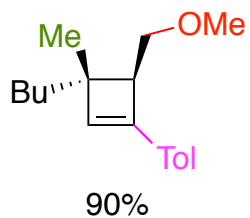
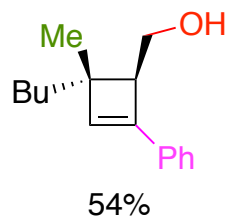
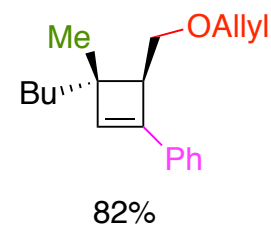
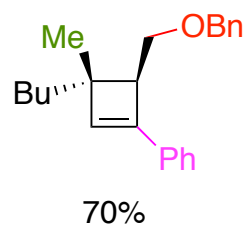
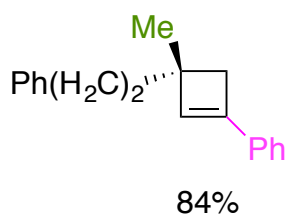
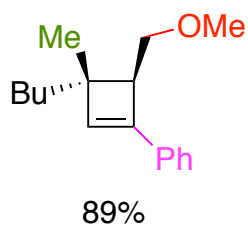
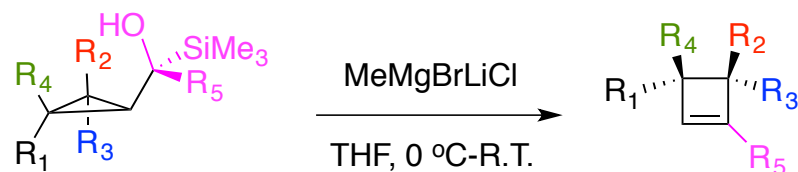
$R_5 = \text{Tol}$; 75%, $dr = 95:5$
 $R_5 = \text{Ph}(\text{CH}_2)_2$; 80%, $dr = 86:14$
 $R_5 = \text{CO}_2\text{tBu}$; 50%, $dr = 66:33$



$R_1 = \text{Bu}$, $R = \text{OH}$, $R_3 = \text{Me}$; 95%, $dr = 98:2$
 $R_1 = \text{Bu}$, $R = \text{OMe}$, $R_3 = \text{Me}$; 77%, $dr = 98:2$
 $R_1 = \text{Bu}$, $R = \text{OMe}$, $R_3 = \text{Ph}$; 72%, $dr = 98:2$
 $R_1 = \text{H}$, $R = \text{OBn}$, $R_3 = \text{Me}$; 45%, $dr = 98:2$
 $R_1 = \text{Bu}$, $R = \text{H}$, $R_3 = \text{Me}$; 40%, $dr = 58:42$

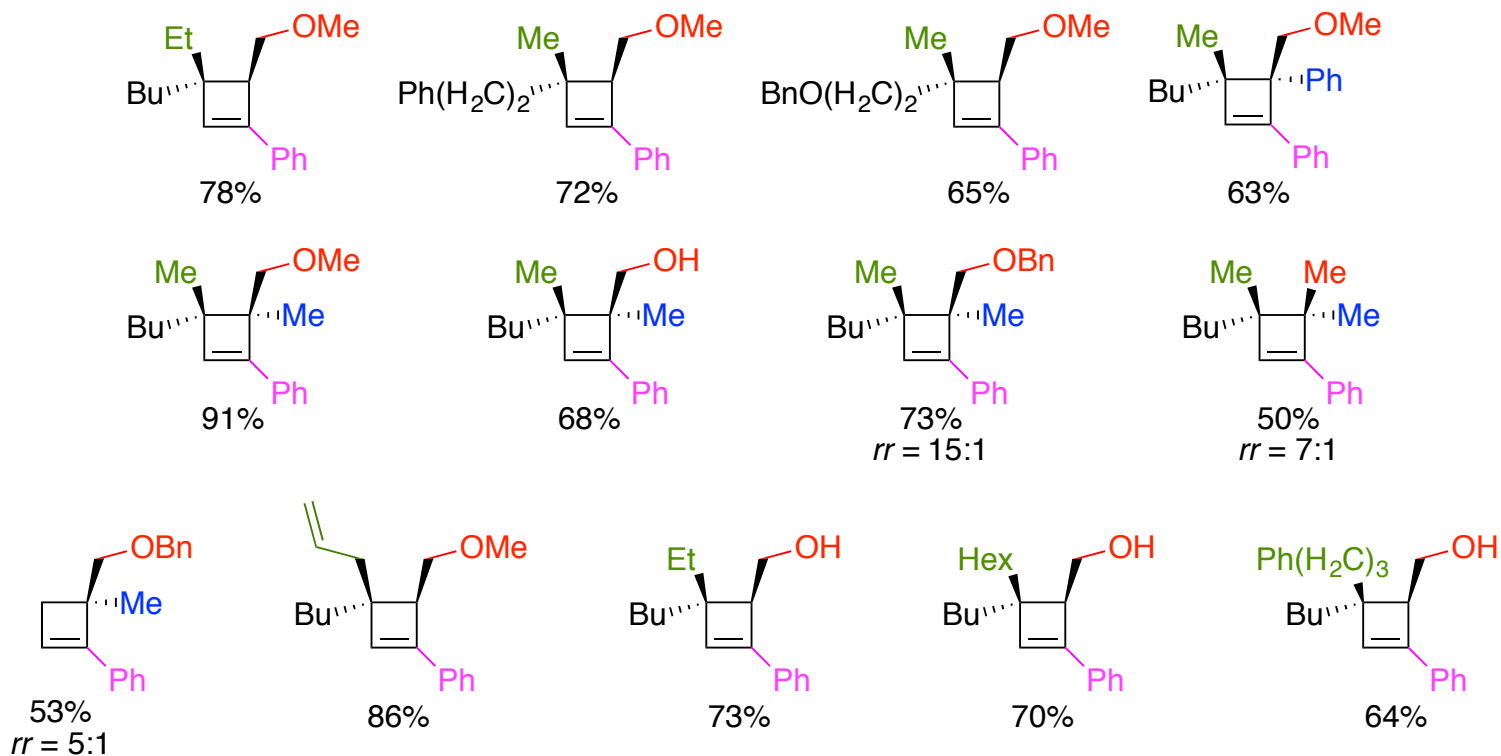
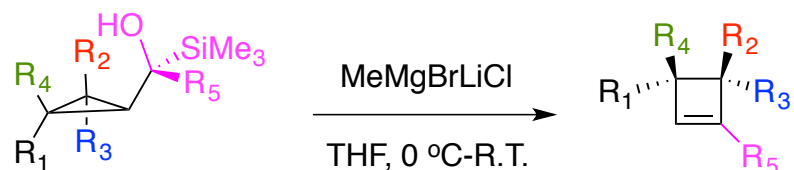
J. Am. Chem. Soc.,
2017, *139*, 8364–
 8370.

Title Work: Ring Expansion



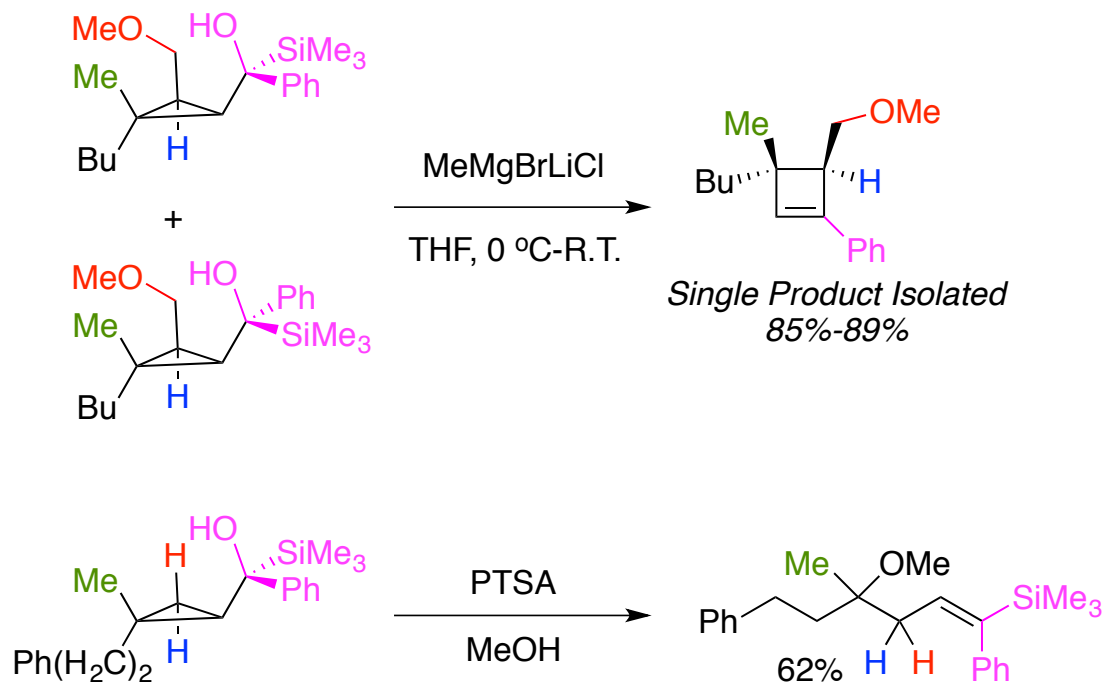
J. Am. Chem. Soc., **2017**, *139*, 8364–8370.

Title Work: Ring Expansion



J. Am. Chem. Soc., 2017, 139, 8364–8370.

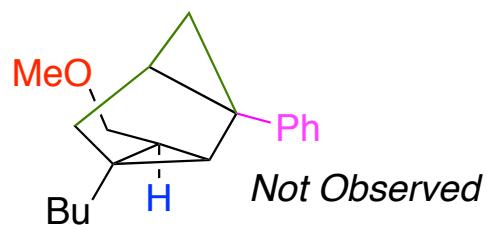
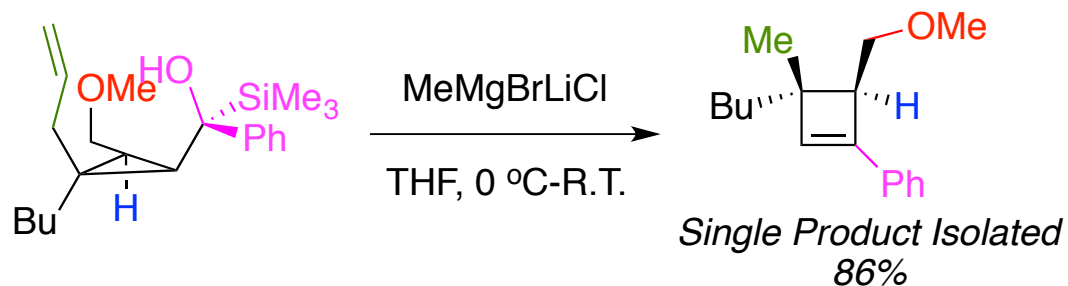
Probing the Mechanism



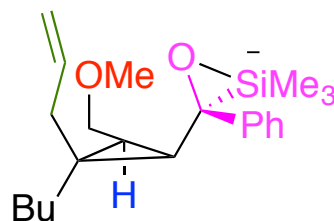
- Different diastereomers give same product
 - Suggests that carbene pathway possible
- PTSA reaction: Test if cationic reaction pathway is operative

J. Am. Chem. Soc., 2017, 139, 8364–8370.

Probing the Mechanism

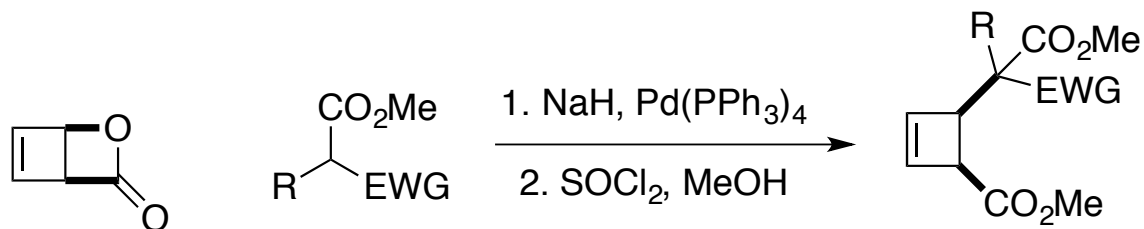
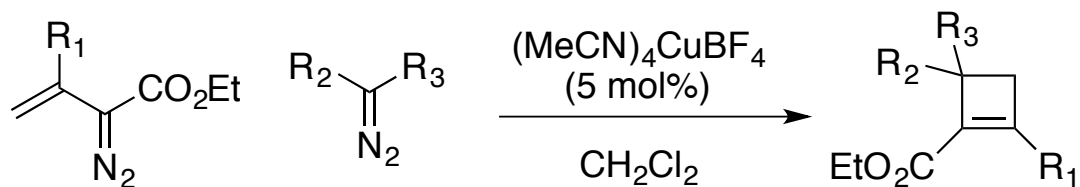
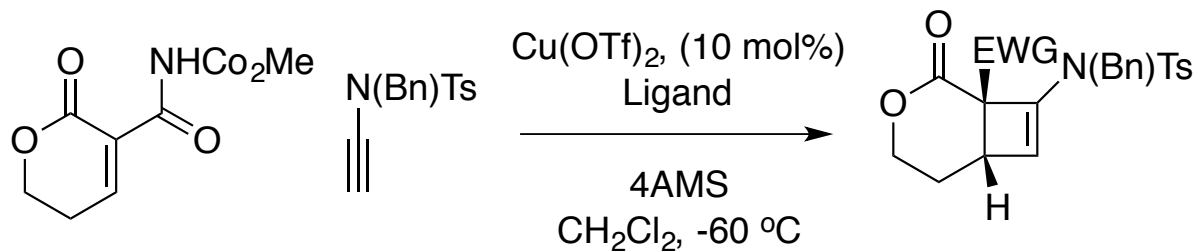


- Attempts to trap carbene fail
 - Analogous to similar previous report
 - Possible alternative?



J. Am. Chem. Soc., **2017**, *139*, 8364–8370.
Org. Bio. Mol. Chem. **2004**, *4*, 3006–3017.

Other Cyclobutene Methods



Org. Lett., **2010**, *17*, 3780-3783.
Angew. Chem. Int. Ed. **2009**, *48*, 7569-7572.
Angew. Chem. Int. Ed. **2010**, *49*, 5762-5766.

Conclusion

- One-pot synthesis of cyclobutenes from cyclopropenes was developed
- Good diastereoselectivity
- Simple protocol